[This question paper contains 4 printed pages.]

Sr. No. of Question Paper: 1611 C Roll No......

Unique Paper Code : 217251

Name of the Course : B.Sc. (Hons.) Biochemistry/Botany/Zoology/BMS

Name of the Paper : Chemistry - II (CHCT -402)

Semester : II

Duration : 3 Hours Maximum Marks : 75

Instructions for Candidates

1. Write your Roll No. on the top immediately on receipt of this question paper.

2. Attempt any FIVE questions.

3. All questions carry equal marks.

1. From the following attempt any FIVE:

(a) Giving explanation, arrange the following in increasing order of C+-C bond strength:

$$H_1C \longrightarrow CH_2$$
, $H_2C \Longrightarrow CH_2$ and $HC \Longrightarrow CH_2$

(b) Which of the following is more basic and why?

$$-CH_2NH_2$$
 and H_3C--NH_2

(c) Explain which of the following will undergo heterolytic cleavage of Carbon-chlorine bond (shown below as thick line) much faster?

$$CH_3$$
— CH_2 — CI and CH_3 — CH — CI
 CH_3

- (d) Explain the increasing order of acidic strength of the following: CH_CH_COOH and BrCH_COOH
- (e) Giving reasons, arrange the following in increasing order of stability (CH₂) CH. CH₂CH₃ and (CH₃) C
- (f) Out of *ortho*-nitrophenol and *para*-nitrophenol which one is more volatile and why? (3×5)
- 2. (a) Assigning priority order, explain how will you designate \mathbf{E} -/ \mathbf{Z} to the following:

$$H_3C$$
 OH HOOC Br

 H_3C H_3C H_4 (4)

(b) Indicate which of the following stereoisomer designated as *threo* and which as *erythro* and why?

$$H - OH H - OH H - OH H - OH CH2OH $CH2OH$ $CH2OH$ (i) $(ii)$$$

(c) Draw different confomations of cyclohexane and arrange them in their decreasing order of stability, giving suitable explanation. (4)

(d) Arrange the following free radicals in decreasing order of stability and explain your answer on the basis of hyperconjugation.

$$CH_3 \dot{C}H_2$$
, $(CH_3)_5 \dot{C}H$ and $(CH_3)_5 \dot{C}$ (5)

- 3. (a) Draw the Fischer projection for all possible steroisomers of 2.3-dihydroxybutane. What is the correlation among these steroisomers? Comment on optical activity of these stereoisomers. (5)
 - (b) Assigning priority order, explain how will you arrive at R-/S- configuration of the following:

$$CH_{3}$$

$$H = -\frac{1}{1} - CI$$

$$CH_{2}OH$$

$$(i)$$

(c) Explaining the steps involved how will you convert the following into Fischer projection:

- 4. (a) What are essential and non-essential amino acids? Give two examples of each. (3)
 - (b) Give the preparation of Glycine using Gabriel phthalimide synthesis. (3)

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	(c)	Explain Edman degradation in determining end group and peptides.	alysis in (3)
	(d)	Give the synthesis of dipeptide Gly-Val from Glycine and Vali Merrifield solid phase peptide synthesis.	ine using (6)
5.	(a)	Carry out the following conversions:	
		(i) D-glucose to D- fructose	
		(ii) D-arabinose to D-glucose	(3,3)
	(b)	How will you establish the open chain structure of glucose (configuration).	ration not (6)
	(c)	Draw the Haworth projection formula for	
		(i) β-D(+)-glueopyranose	
		(ii) α-D()-fructofuranose	(2)
	(d)	Name the monosaccharides obtained on hydrolysis of sucrose.	(1)
6.	Write a short note on any THREE of the following:		
	(i)	Mutarotation	
	(ii)	Flectrophoresis for separation of amino acids	
	(iii)	Primary, secondary and tertiary structure of proteins	
	(iv)	Ruff's degradation in carbohydrates	(5,5,5)