

[This question paper contains 2 printed pages.]

Sr. No. of Question Paper : 8591

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Roll No.....

Unique Paper Code : 217573

Name of the Paper : GEHT-504 : Inorganic Chemistry – I

Name of the Course : B.Sc. (H) Geology, Part III

Semester : V

Duration : 3 Hours

Maximum Marks : 75

Instructions for Candidates

1. Write your Roll No. on the top immediately on receipt of this question paper.
2. Attempt any **five** questions.
3. Each question carry **15** marks.

1. (a) Derive the following forms of Gibbs – Duhem Equation.

$$Sdt - Vdp + \sum n_i d\mu_i = 0 \text{ (where } i=1 \text{ to } n.)$$

- (b) Define partial molar quantity.

- (c) Show that the variation of chemical potential of a component 'i' with pressure is given by

$$d\mu_i = V_{i,m} dp$$

2. (a) Derive an expression for the free energy of mixing when n_1 moles of nitrogen and n_2 moles of H_2 are mixed isothermally and isobarically.

- (b) Show that the molar free energy of mixing ΔG_{mix} in a binary ideal gas mixture is minimum when two gases are present in the equimolar ratio.

- (c) Calculate free energy of mixing ΔG_{mix} at $25^\circ C$ and 1 atm when 10 moles of He are mixed with 10 moles of Ne.

3. (a) Write the role of Na^+ and K^+ ions in the living system.

- (b) Explain the role of Metal Chelates in living system.

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- (c) Write the structure of Chlorophyll.
4. (a) Give two methods of preparation of diborane and discuss its structure.
- (b) Explain the thermal stability of the hydrides of group I and group II elements.
- (c) What happens when NaH reacts with
- (i) CO
 - (ii) SiCl_4 .
 - (iii) Fe_3O_4 .
5. (a) Define the following :
- (i) Collision number.
 - (ii) Collision diameter.
 - (iii) Mean free path of molecules.
 - (iv) Collision frequency.
 - (v) Degrees of freedom of motion.
- (b) What is the effect of temperature and pressure on the coefficient of viscosity.
- (c) Define the principle of equipartition of energy.
6. (a) What is the effect of temperature and pressure on collision frequency.
- (b) Define viscosity. Give its S.I. unit.
- (c) What is the effect of temperature on viscosity of a liquid and gases.?
- (d) Define Surface Tension. Give its S.I. unit.
- (e) What is the effect of temperature and pressure on surface tension of a liquid ?
- (f) Why drop number method is more accurate than drop weight method for the determination of surface tension of a liquid by Stalagmometer ?