[This question paper contains 5 printed pages.]

4718

Your Roll No.

B.Sc.(G) / I

AS

CHEMISTRY - Paper II

(Organic Chemistry)

Time: 2 hours

Maximum Marks: 25

(Write your Roll No. on the top immediately on receipt of this question paper.)

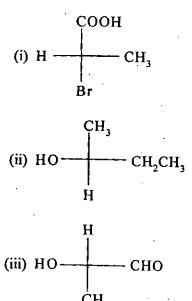
Attempt any four questions.

- (a) What is hybridization? What are distinguishing features of sp³ and sp² orbitals?
 - (b) Indicate the type of hybridization of each carbon atom in the following structures:
 - (i) CH_3CH_3 (ii) $CH_3C \equiv CCH_3$ (iii) CH_3OH
 - (c) Give IUPAC systematic name for each of the following:

P.T.O.

- (d) A sample containing 1.00 gmL⁻¹ of optically active substance, in a 1-dm tube, produced a rotation of 3.12° in a clockwise direction, at 25°C and D line of sodium lamp. Calculate the specific rotation?

 (2,1½,1,1½)
- (a) Assign R or S configurations to the chirality centers in the following molecules:



(b) Assign E or Z configuration to each of the following compounds:

(i)
$$C = C$$
 $CH = CH_2$ CH_3 CH_3

CH₃
$$CH_2CH(CH_3)_2$$
(ii) $C = C$

$$CH_3$$

- (c) What are enantiomers and diastereomers? Explain with suitable examples.
- (d) Dehydration of 2-Methyl-2-butanol with acid gives a mixture of two butenes. Write out the chemical equation, state which product predominates and give reasons. (1½,1½,2,2)
- (a) Give the mechanism and synthesis of Ethyl-2,
 2-dimethylpropyl ether, using Williamson's synthesis.
 - (b) Identify the products A to F in the following reactions:

(i)
$$(CH_3)_2C = 0 + 2:4$$
-Dinitrophenyl hydrazine-

(ii)
$$O + NH_2OH \longrightarrow B$$

$$\begin{array}{c|cccc}
H & & H^+ & ROH, H^+ \\
(iv) R' - C = O + ROH & & E & & F
\end{array}$$

- (c) Write down the structure of a carbanion. Discuss about its stability.
- (d) Describe Lucas test to distinguish between primary, secondary and tertiary alcohols.

(1,3,1,1)

- (a) Write down the basic principle involved in the chromatographic purification of organic compounds.
 - (b) (i) Carboxylic acids are much more acidic than the corresponding alcohols. Explain.
 - (ii) Draw resonance forms of nitromethane.
 - (c) Give the mechanism and Kinetics of S_N² reaction, between methylbromide and hydroxide ion. Write down the order of reactivity of primary, secondary and tertiary alkyl halides.
 - (d) (i) Give a method for the preparation of Ethanethiol.
 - (ii) How can dichlorocarbene be generated?
 (1,2,1½,1½)

- (a) Give the mechanism and preparation of Methylamine using Hofmann's rearrangement.
 - (b) (i) Primary aliphatic amines are stronger bases than ammonia. Explain.
 - (ii) Discuss the mechanism of Aldol condensation.
 - (c) Rank the substances in each of the following groups in order of increasing acidity:
 - (i) CH₃CH₂COOH, BrCH₂COOH, FCH₂COOH
 - (ii) CH₃CH₂OH, CH₃CH₂NH₂, CH₃CH₂COOH
 - (d) Write a note on synthetic uses of grignard reagents. (1½,2,1,1½)