

[This question paper contains 2 printed pages.]

4719

Your Roll No. ....

B.Sc. (G) / I

AS

CHEMISTRY – Paper III

(Physical Chemistry)

Time : 2 Hours

Maximum Marks : 25

*(Write your Roll No. on the top immediately  
on receipt of this question paper.)*

*Attempt any four questions.*

*Use of Calculator is allowed.*

1. (a) In a measurement of viscosity by Ostwald's viscometer at  $20.2^{\circ}\text{C}$ , water takes 30 seconds to flow between the upper and lower marks while the flow of another liquid of density  $1.5\text{ g/c.c.}$  takes 35 seconds. Taking density of water at  $20.2^{\circ}\text{C}$  to be  $0.9982\text{ g/c.c.}$ , calculate the viscosity of the other liquid. (Viscosity of water at this temperature is 10 centipoise.) (3)
- (b) Calculate the r.m.s. velocity of oxygen molecules at  $27^{\circ}\text{C}$ . (2)
- (c) The resistance of a N/10 solution is found to be  $2.5 \times 10^3\text{ ohms}$ . Calculate the equivalent conductance of the solution if the all constant =  $1.15\text{ cm}^{-1}$ . (2)

P.T.O.

2. (a) Explain the term co-efficient of viscosity. (2)  
(b) Describe Ostwald's viscometer method for the determination of viscosity of a liquid. (4)
3. (a) What is meant by transport number of an ion? (2)  
(b) Describe moving boundary method for the determination of transport numbers. (4)
4. (a) What is a buffer solution? (2)  
(b) Derive Henderson equation for the pH of an acidic buffer mixture. (4)
5. Write short notes on any two :
- (a) Law of rational indices
- (b) Kammerlingh-Dunes virial equation of state for real gases
- (c) Law of Equipartition of Energy (2×3)