This question paper contains 4 printed pages;

Your Roll No.

209

B.Sc. (Prog.)/II

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-CH-202-ORGANIC CHEMISTRY

(Admissions of 2008 and onwards)

Time: 2 Hours Maximum Marks: 50

True your Read of the transport methods on the great most question paper is

Attempt any four questions

- (a) What happens when a quaternary ammonium hydroxide is heated strongly? Explain with the help of an example.
 - (b) What products would you obtain from the KMnO₄ oxidation of the following compounds?

$$(i) \qquad O_2N \qquad \qquad CH(CH_3)_2$$

(c) HBr can add to propene under different reaction conditions to give isomeric products. Explain. 41/2,4,4

2. (a)
$$CH_3CH=CH_2 \longrightarrow A \xrightarrow{CH_3CH=CH_2} H_-^+BH_2$$

$$(CH_3CH_2CH_2)_2BH \xrightarrow{CH_3CH=CH_2} B$$

Write the structures of A and B.

(b) Complete the following reactions:

(i)
$$R-C = C-R' \xrightarrow{(I) O_3} ?$$

(ii)
$$CH_3CH=CH_2 \xrightarrow{\text{(I) OsO}_4, \text{ pyridine}}$$
?

- (c) (i) Give the structure of the products when acetaldehydeis reacted with :
 - (1) H₂NOH
 - (2) C₂H₅OH
 - (ii) Write down the mechanism of the reaction of a primary alcohol with SOCI₂. 4,4,4½

- 3. (a) (i) In most Friedel-Crafts acylation the electrophile is an acylium ion. How it can be formed from an acylhalide?
 - (ii) Give important limitations of Friedel-Crafts reactions.
 - (b) How would you prepare 2-pentanone by an acetoacetic ester synthesis?
 - (c) Complete the following reaction: $4\frac{1}{2}.4.4$

$$\begin{array}{c} \overset{O}{\text{RCH}} \xrightarrow{\text{KMnO}_4, \text{ OH}^-} ? & \xrightarrow{\text{H}_3\text{O}^+} ? \\ \xrightarrow{\text{Ag(NH}_3)_2^+} ? & \xrightarrow{\text{H}_2\text{O}} ? \end{array}$$

- 4. Write notes on the following: $4^{6} \pm 4.4$
 - (i) Pinacol-Pinacolone rearrangement
 - (ii) Benzidine rearrangement
 - (iii) Perkin reaction

- 5. (a) (i) What products are obtained when nitrobenzene is reduced in acidic and alkaline mediums, respectively.
 - (ii) How would you prepare the following compounds starting from toluene :
 - (1) m-toluidine
 - (2) *m*-bromotoluene.

(ii)
$$C-CH_2CH_2CH_3 \xrightarrow{Zn(Hg)} A$$

$$C-CH_2CH_2CH_3 \xrightarrow{Conc. HCl} A$$

$$C-CH_3 \xrightarrow{LiAlH_4} B$$

Identify A and B.

(c) Give the mechanism of haloform reaction. What is its synthetic utility?