[This question paper contains 3 printed pages.]

4601

Your Roll No.

B.Sc. Prog./II

AS

CH-202: ORGANIC CHEMISTRY

(Admissions of 2008 & onwards)

Time: 2 Hours Maximum Marks: 50

(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt any four questions.

Question No. 1 is compulsory and carries 14 marks.

- 1. (a) An aromatic hydrocarbon (A), C_9H_{12} can be oxidized with KMnO₄ to (B) having molecular formula $C_8H_6O_4$ Compound (B) is a dicarboxylic acid but does not form an anhydride on heating (B) when treated with bromine in presence of FeCl₃ gives only one monobromoderivative (C), $C_8H_5BrO_4$. Give the structural formulae for (A), (B) and (C).
 - (b) Identify the compounds (A) through (D) -

(A)
$$\xrightarrow{\text{HNO}_3}$$
 $\xrightarrow{\text{O}}$ $\xrightarrow{\text{Sn}}$ (B) $\xrightarrow{\text{NaNO}_2}$ (C) $\xrightarrow{\text{HCI}}$ (C) $\xrightarrow{\text{O}-5^{\circ}\text{C}}$ $\xrightarrow{\text{O}}$ $\xrightarrow{\text{CH}_3}$ $\xrightarrow{\text{O}}$ $\xrightarrow{\text{Br}}$ P.T.O.

- (c) How will you obtain 1-propanol and 2-propanol from propene. (6,4,4)
- 2. (a) Give the synthesis of ethylacetoacetate from ethylacetate. Give the name and mechanism of the reaction involved.
 - (b) Starting from EAA, how will you prepare -
 - (i) 4-methyl-uracil
 - (ii) Adipic acid.
 - (iii) Pentan -2.4 dione (6.6)
- (a) Giving reasons, arrange the following in the decreasing order of reactivity towards addition of HCl ethene, propene, 2-methylbut-2-ene.
 - (b) Give the products of reduction of nitrobenzene in acidic and electrolytic conditions.
 - (c) Give a chemical test to distinguish between the two of following:
 - (i) Acetone from diethyl ketone
 - (ii) Acetone from acetaldehyde (4×3)
- 4. Explain the following:-
 - (a) Ph-NH₂ does not undergo Friedel crafts alkylation reaction.

- (b) Chlorobenzene has a C-Cl bond length of 1.69 A° while ethylchloride has it 1.76 A°.
 - (c) Halogens are ortho-para directors but are deactivating. (4×3)
- 5. Give the major products obtained when (any six) -
 - (i) Glycerol is treated with periodic acid.
 - (ii) 2-bromobutane is treated with alcoholic KOH.
 - (iii) Benzenesulphonic acid is fused with NaOH.
 - (iv) H₃CCOCI is treated with water/acid.
 - (v) undergoes reductive ozonolysis.
 - (vi) Propyne is treated with H₂O in presence of H₂SO₄/HgSO₄.
 - (vii) H₃CCONH₂ is treated with Br₂ in presence of aq KOH.
 - (viii) t-butylmethylether is treated with HI. (2×6)
- (a) Both o-bromoanisole and m-bromoanisole yield the same product on treatment of KNH₂ in presence of liquid ammonia. Explain giving mechanism.
 - (b) Compare SN¹ and SN² reaction mechanisms involving alkyl halides. (6,6)

(300)****