[This question paper contains 4 printed pages.]

Sr. No. of Question Paper: 7971 F-2 Your Roll No.....

Unique Paper Code : 1141201

Name of the Course : B.Tech. Polymer Science

Name of the Paper : Polymer Chemistry (DC-1.3)

Semester : II

Duration: 3 Hours Maximum Marks: 75

## **Instructions for Candidates**

1. Write your Roll No. on the top immediately on receipt of this question paper.

- 2. Attempt five questions in all.
- 3. Question No. 1 is compulsory.
- 4. Use of calculator is allowed and log table may be provided.
- 1. (a) Are the following statements true or false? Justify your answer
  - (i) Same type of in initiator are used in suspension and emulsion polymerization of vinyl chloride.
  - (ii) Poly( vinyl alcohol) is prepared by free radical polymerization of vinyl alcohol.
  - (iii) Broad molecular weight distribution is observed in anionic polymerization alkyl methacrylates.
  - (iv) Vinyl acetate and vinyl chloride are the most favored copolymerization components for vinyl ethers, but both of them reluctant to copolymerize with styerene.
  - (v) Maleic anhydride does not homopolymerize but react with styrene to form alternating copolymer. (1,1,1,2,1)

- (b) Show chemical crosslinking reactions for cast elastomers of Polyurethane.
- (c) A ethylene polymerization process gives no ethyl or butyl braches and methyl groups upto 3/1000 carbons in polymer. Discuss the said process.
- (d) In spite of similarities with HDPE, i-PP differs in many respects. Explain.
- (e) Write equations for *CABAC* block copolymer synthesis. where A, B, and C represent styrene, butadiene, and isoprene, respectively.
- (f) Estimate the feed and copolymer compositions for the azeotropic copolymerization of Acrylonitrile  $(r_1 = 0.04)$  and styrene  $(r_2 = 0.4)$  at  $60^{\circ}$ C.
- (g) Contrary to the high selectivity shown in cationic and anionic polymerization, radical initiators may bring about polymerization of any carbon-carbon double bond. Explain, giving reasons.
- (h) For emulsion polymerization it is assumed that spoke like free radicals produces in aqueous enter micelles rather than emulsified monomer droplets. How would justify this assumption? (7×3)
- 2. (a) Derive the expression for polymerization rate for an anionic polymerization where termination occurs simultaneously with propagation.
  - (b) Discuss Sioplas process for crosslinking of Polyethylene.
  - (c) Illustrate the low stereo selectivity of  $\beta$ -TCl<sub>3</sub> in Ziegler Natta polymerization. (5,4,3)
- 3. (a) The following data were obtained for the polymerization of methyl methacrylate in benzene at 77°C with azo-bis-isobutyronitrile initiator. Assuming that the initiator efficiency is independent of monomer concentration, are the data consistent with the model for the rate of polymerization by free radical mechanism.

[M] (kmol/m³)	[I] (mol/m³)	−d[M]/dt (mol/m³ · sec)
9.04	0.235	0.193
8.63	0.206	0.170
7.19	- 0.255	0.165
6.13	0.228	0.129
4.96	0.313	0.122
4.75	0.192	0.0937
4.22	0.230	0.0867
4.17	0.581	0.130
3.26	0.245	0.0715
2.07	0.211	0.415

- (b) Discuss properties of silicon fluids.
- (c) What are Telechelic polymers? How are these prepared? (5,4,3)
- 4. (a) Write chemical equations for the polymerization of the following:
  - (i) Acrylonitrile by NaOCH,
  - (ii) Isobutylene by BF<sub>3</sub>/H<sub>2</sub>O
  - (b) Discuss relative physical properties of impact polystyrene, ABS resins and polystyrene.
  - (c) Discuss modification of epoxies with amino resins with suitable reactions. (5,4,3)
- 5. (a) Explain by equations the polymerization of phenol formaldehyde (resol) to form a crosslinked structure.
  - (b) For a chain growth polymerization of a monomer, the following observations were made in chain transfer reaction

- Molecular weight becomes broader &
- · No of polymer molecules remain same

Suggest name of monomers and give reaction for such transfer reaction.

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- (c) The preparation of poly(ethylene terephthalate) from terephthalic acid and ethylene glycol is stopped at 99% conversion. Calculate:
  - (i) The number-average degree of polymerization.
  - (ii) The weight-average degree of polymerization. (5,4,3)
- 6. (a) Discuss the thermal degradation behavior of following with appropriate reactions:
  - (i) Polystyrene
  - (ii) PVC
  - (b) Suggest methods for control of molecular weight in Zeigler Natta polymerization.
  - (c) Complete the following reactions

$$NH_2CONH_2 + base \longrightarrow A + B$$

$$A + CH_2O \longrightarrow C$$

$$C + B \longrightarrow D + base \qquad (5,4,3)$$